

Energieopslag in brandstoffen:

processen en procesintegratie

uitdagingen



Wiebren de Jong (3mE, Process & Energy)

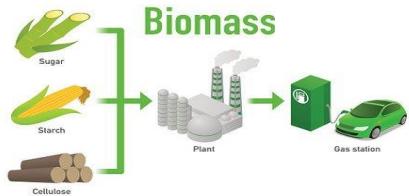
wiebren.dejong@tudelft.nl





Large Scale Energy Storage section

'Biomass route'



Prof. dr. Wiebren de Jong



'Indirect route'

Renewable power based

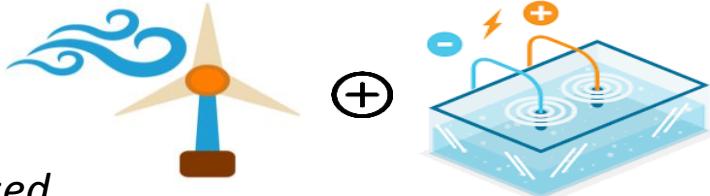


Dr. Wim Huijse



'Direct route'

Renewable power based

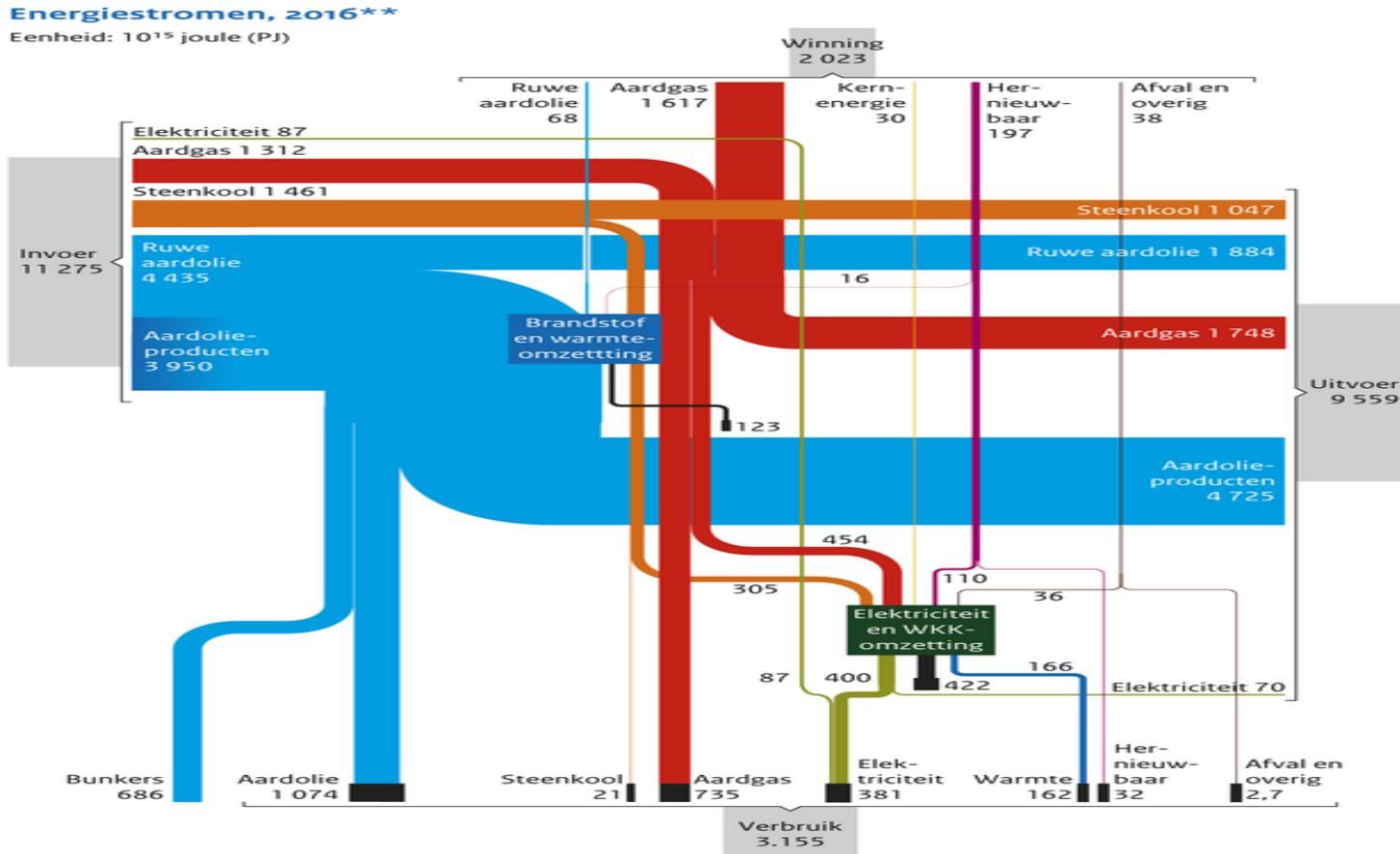


Dr. Ruud Kortlever





Dutch energy supply, mainly fossil based





Wind/solar energy supply is intermittent...

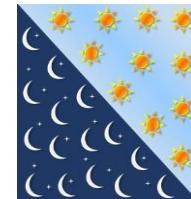


Timescales

<~15-30 min



~12 h



days

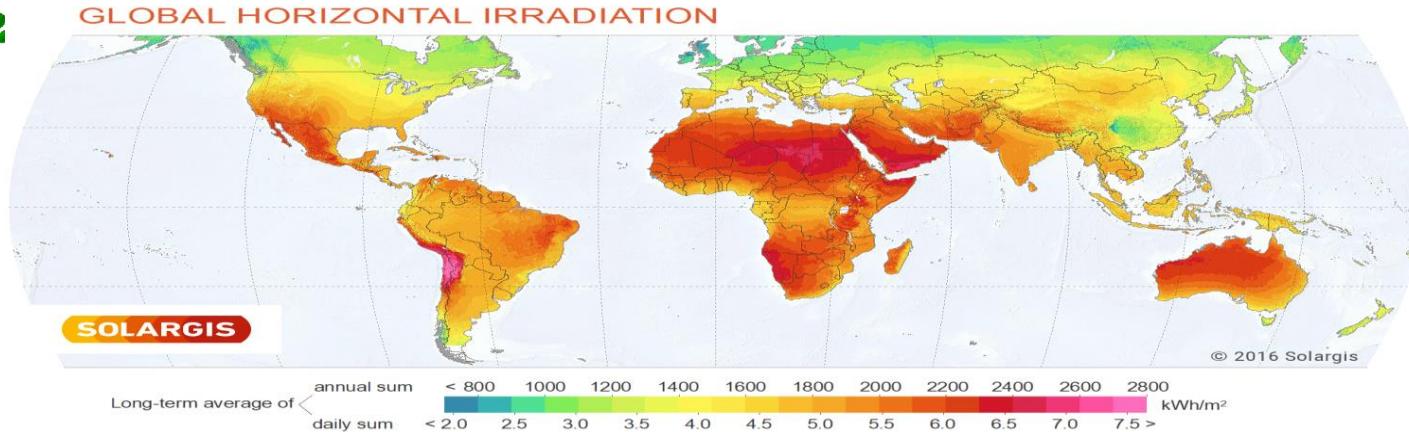


weeks-
months



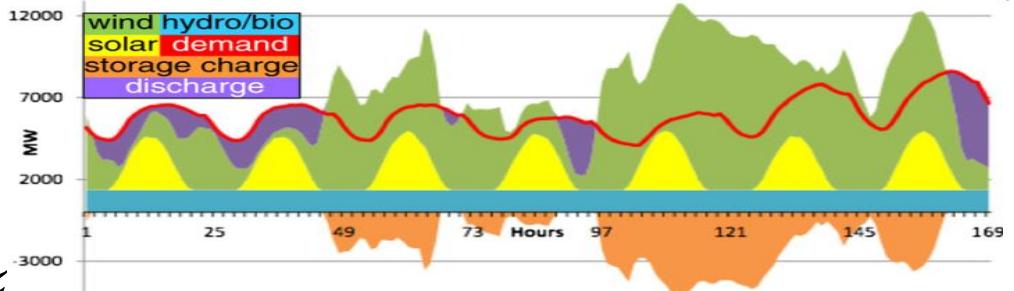
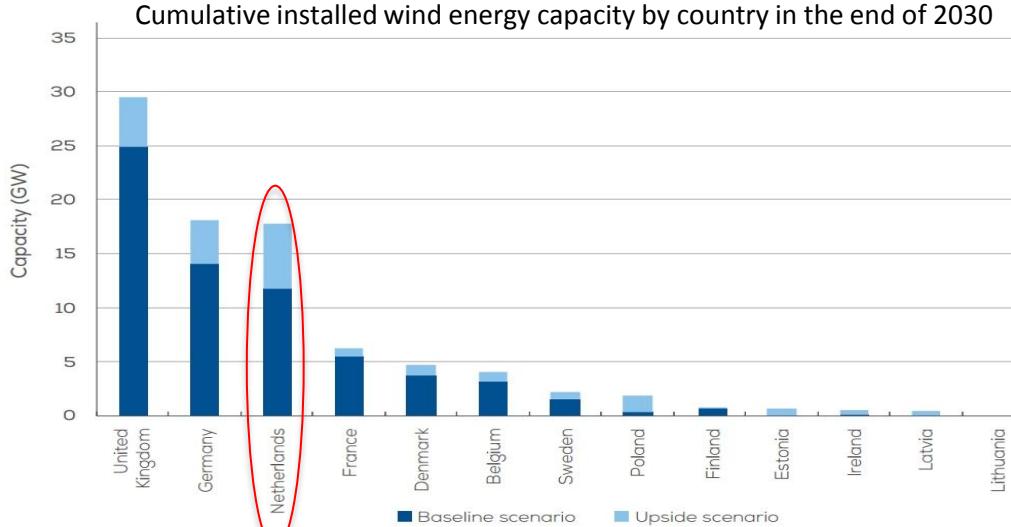
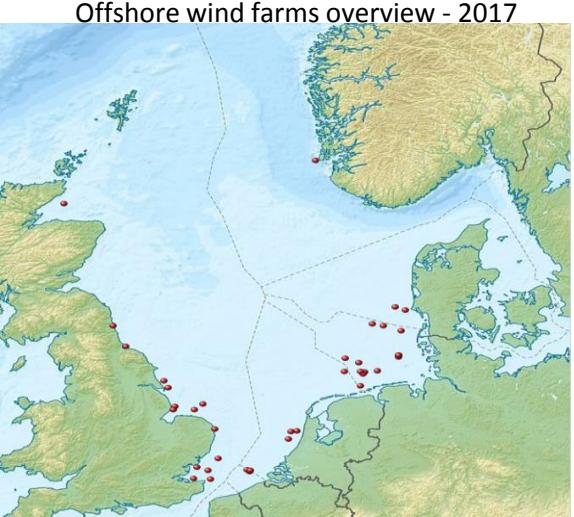


Geographic mismatch demand/supply



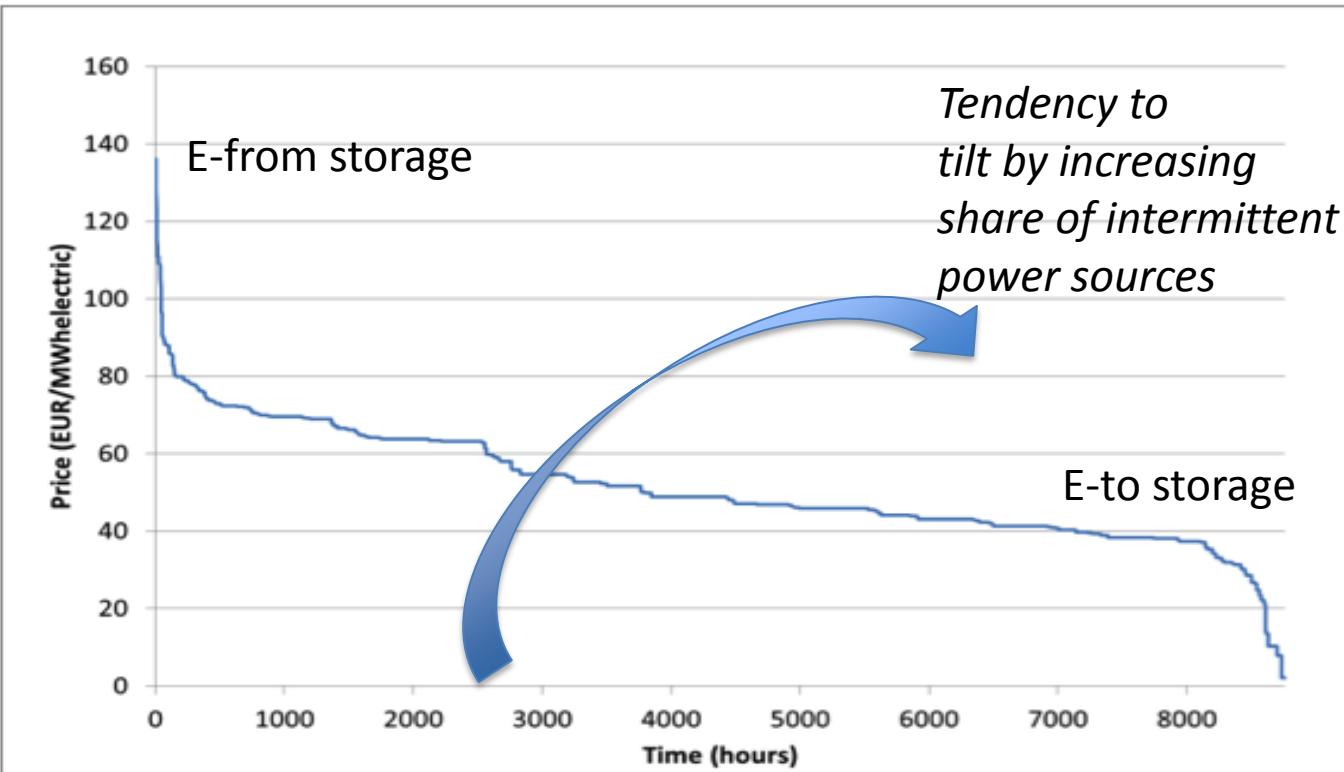


Need for energy storage, NL context





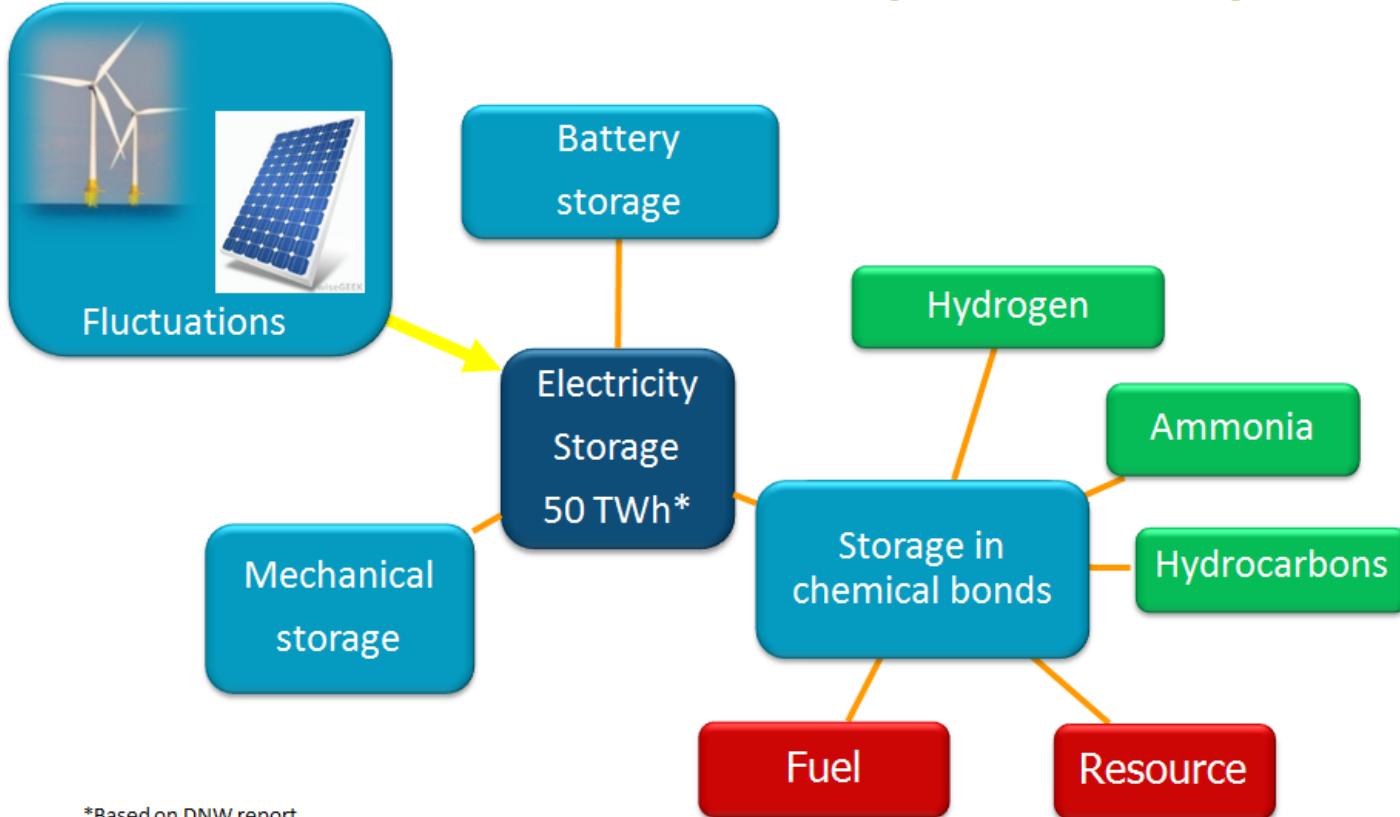
Possible scenario for power price-duration curve 2030



Source (adapted): <http://www.voltachem.com/news/when-why-power-2-heat>



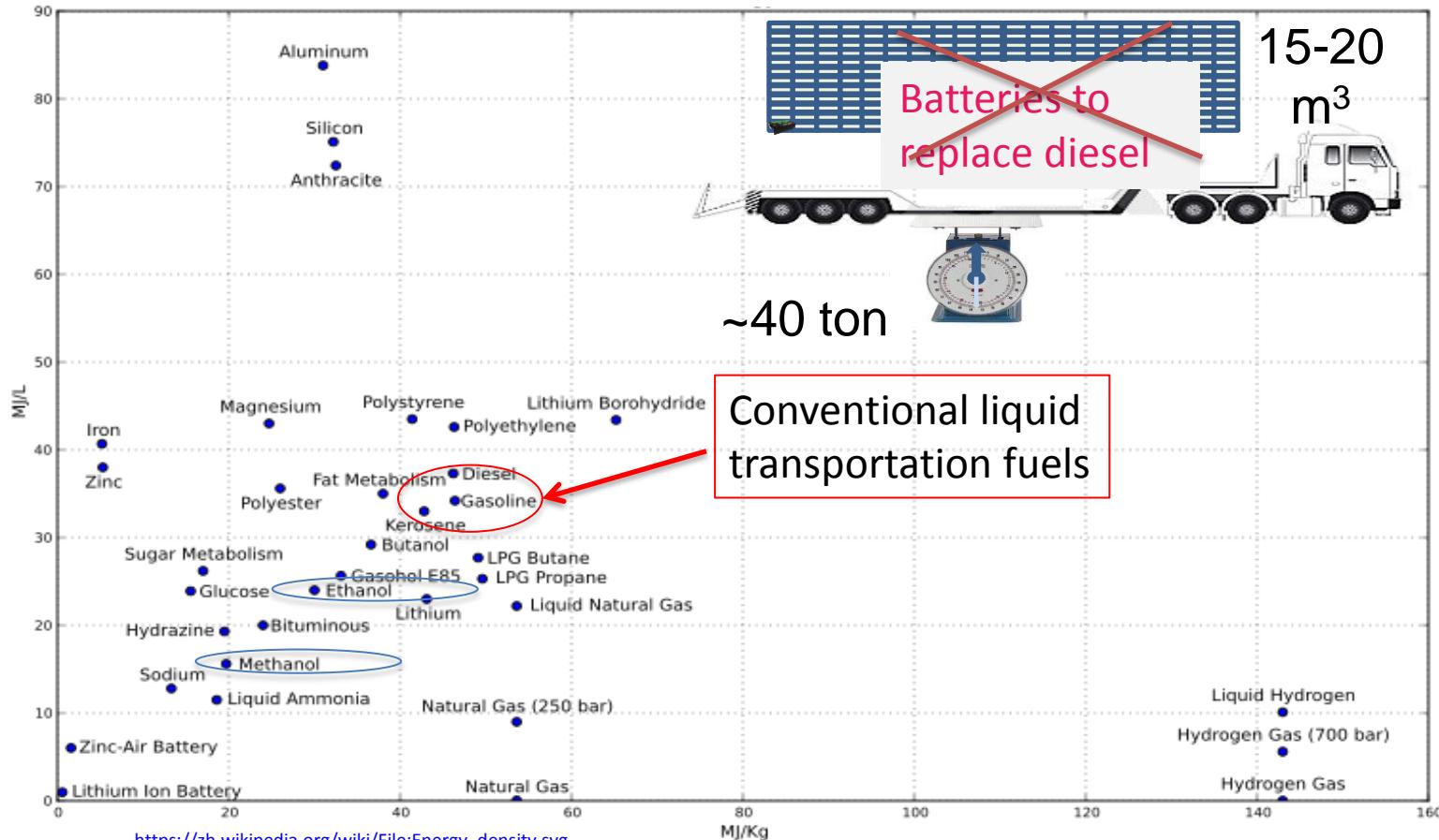
Modes of energy storage



*Based on DNW report



Energy density of several fuels

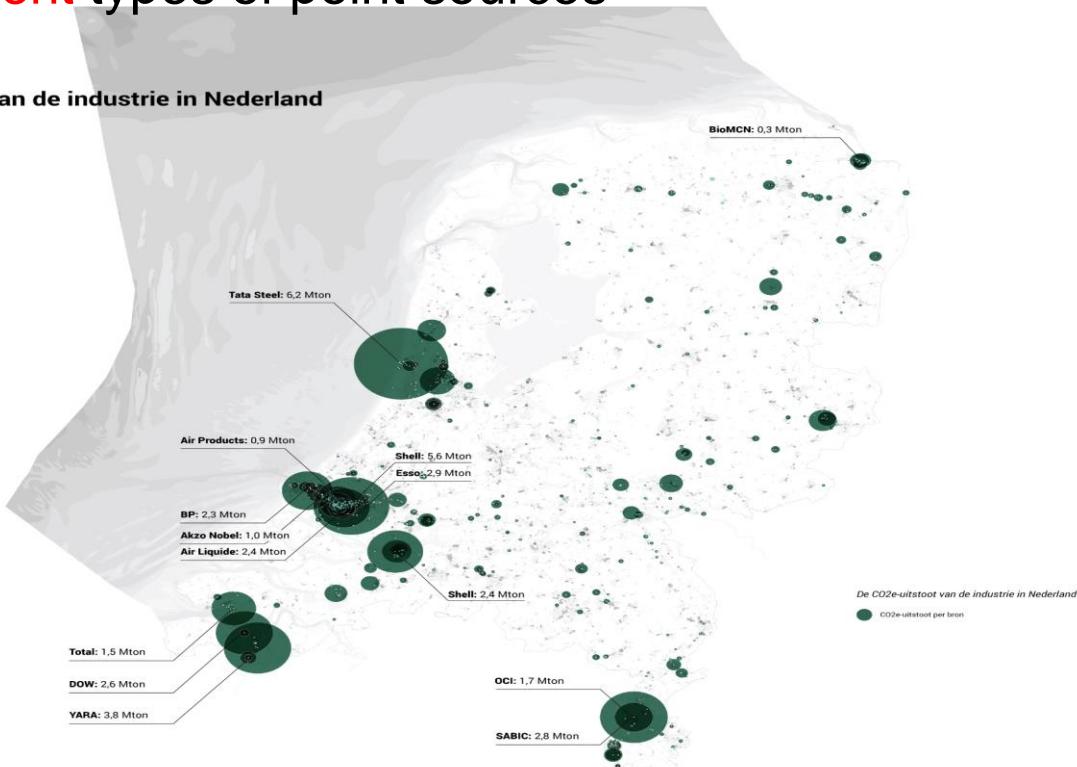




Dutch industry CO₂ reduction

- Reduction by 14.3+5.1 Mton/yr (2030) from 55.1 Mton/yr (2015)
- All **different** types of point sources

CO₂e-uitstoot van de industrie in Nederland

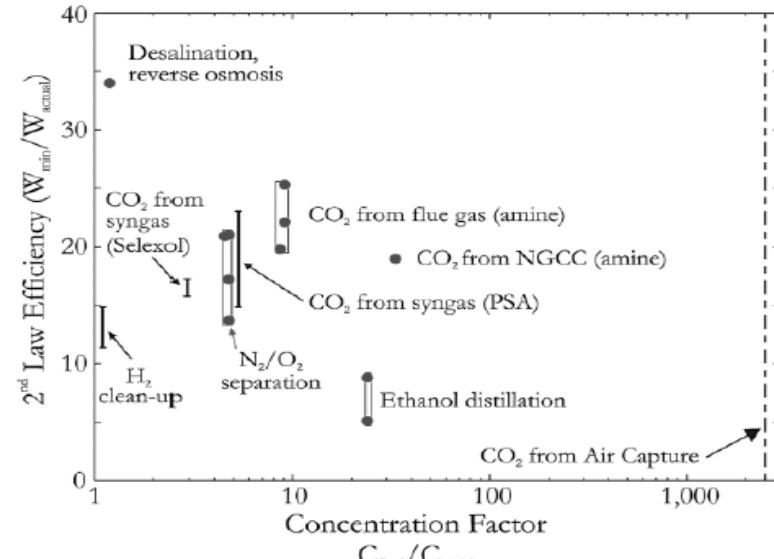
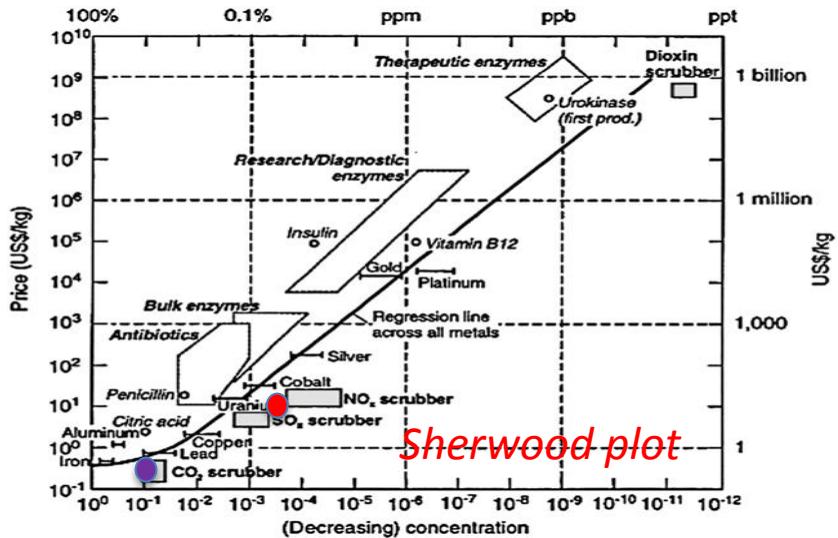


CO₂ sources



Advantageous to use concentrated CO₂ streams *first*

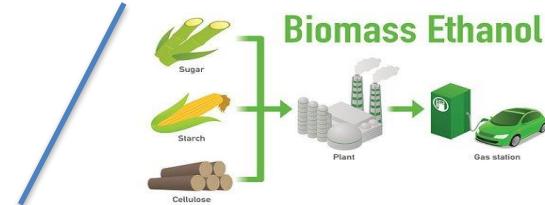
- Minimum work increases when more dilute $\Delta G^\circ = -RT_o \ln y_{\text{CO}_2} \approx 0.5 \text{ MJ.kg}^{-1}$
- High 2nd Law of thermodynamics efficiency
- Costs increase substantially for decreasing concentration



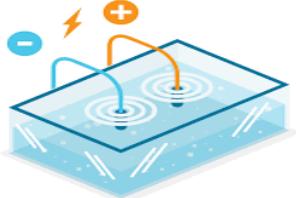


Scenario's for future production of chemicals and fuels

'Conventional'



'Direct route'
Renewable power based



e-Refinery

'Indirect route'

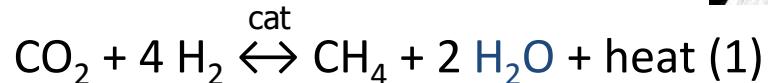
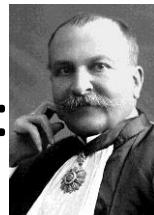
Renewable power based



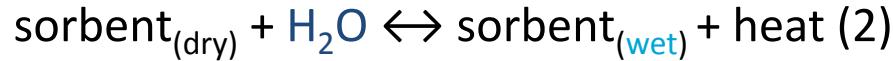


Indirect power-to-X route to methane

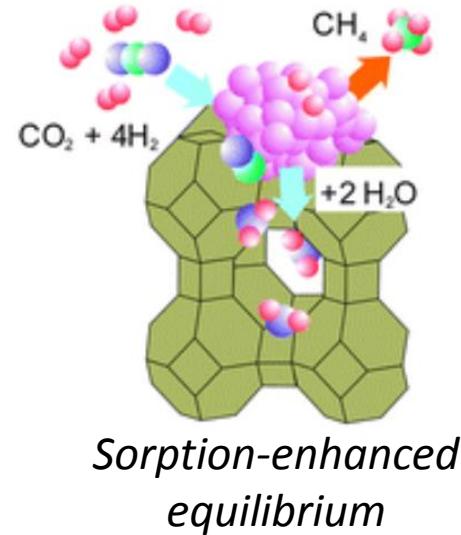
- Sabatier reaction (1902!):



- Shift of the equilibrium by sorption:

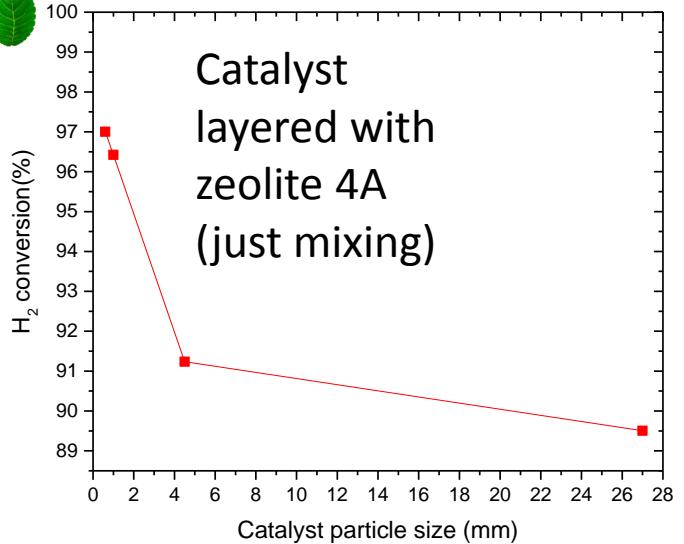


- Dry (=regenerate) sorbent with heat (1)

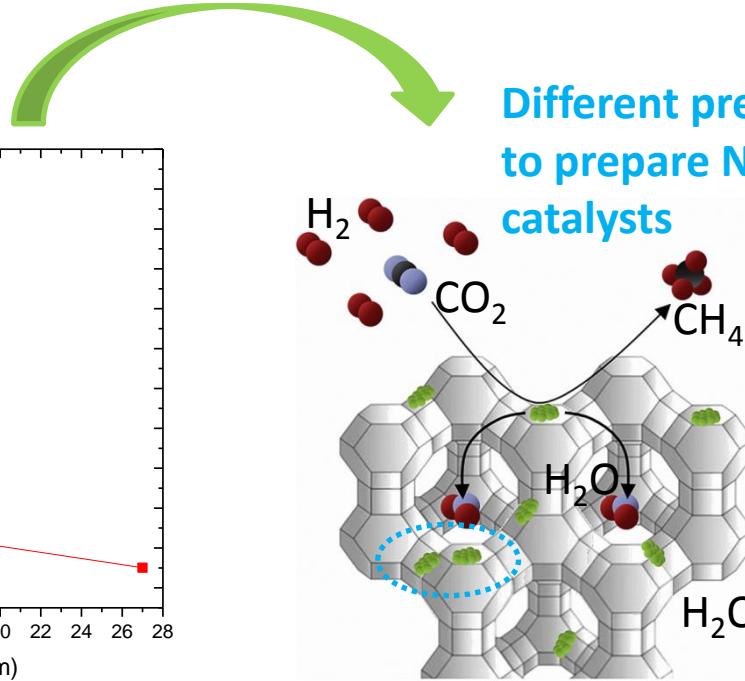




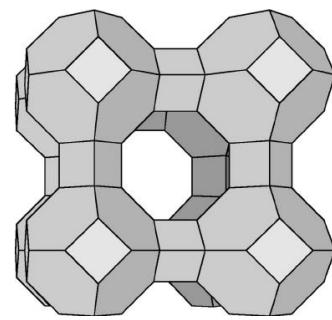
Research towards new catalysts



H₂ conversion under the condition of different size catalyst(42g) mixed with zeolite 4A (2.5-5mm, 220 g) at 280 °C, GHSV=2400 h⁻¹.



Zeolite 13X*

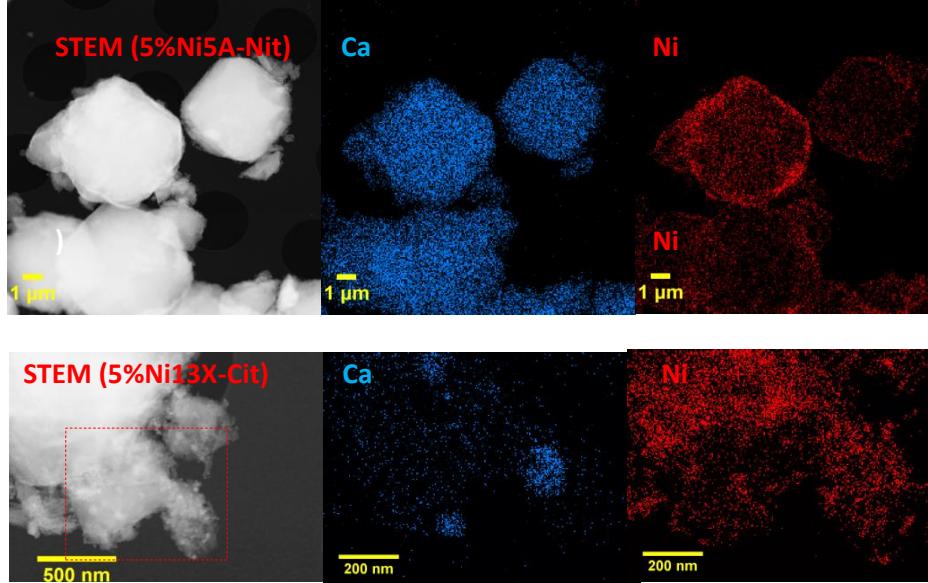


Zeolite 5A

Sorption enhanced CO₂ methanation over a nickel catalyst on zeolite

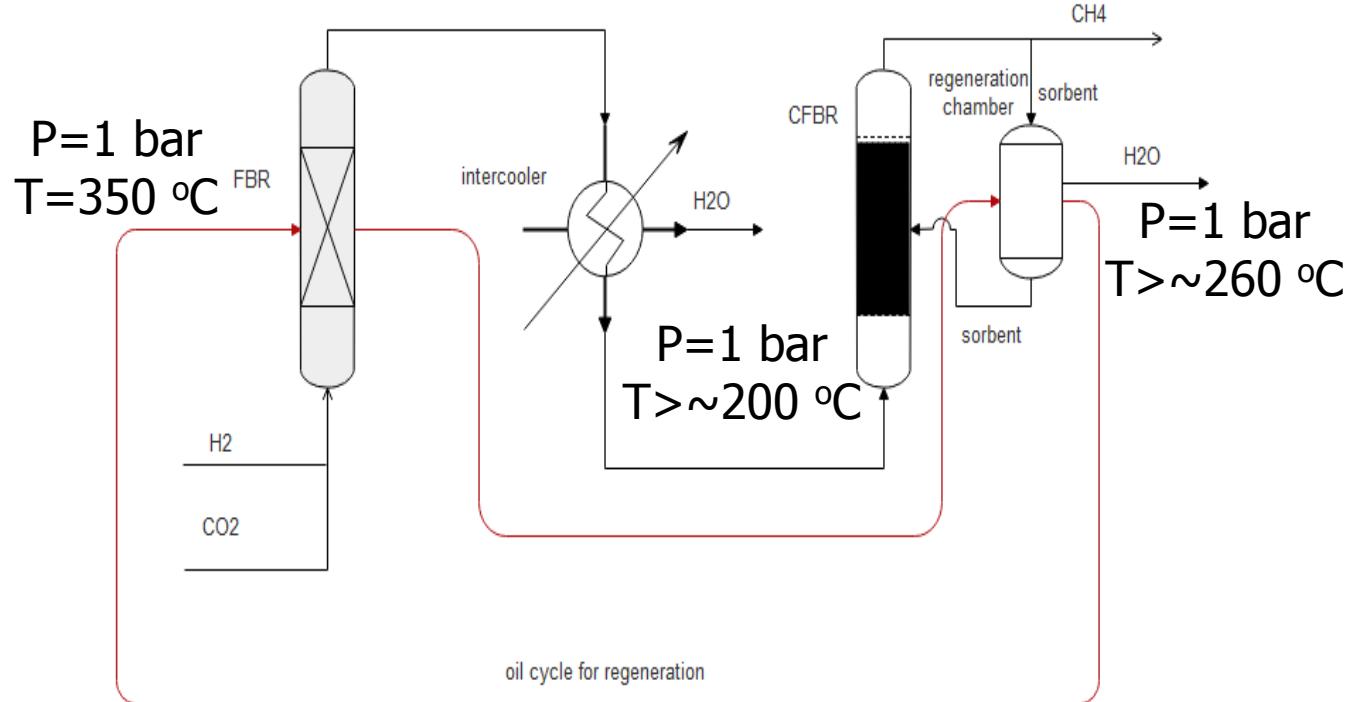


Material preparation, impact on Ni distribution



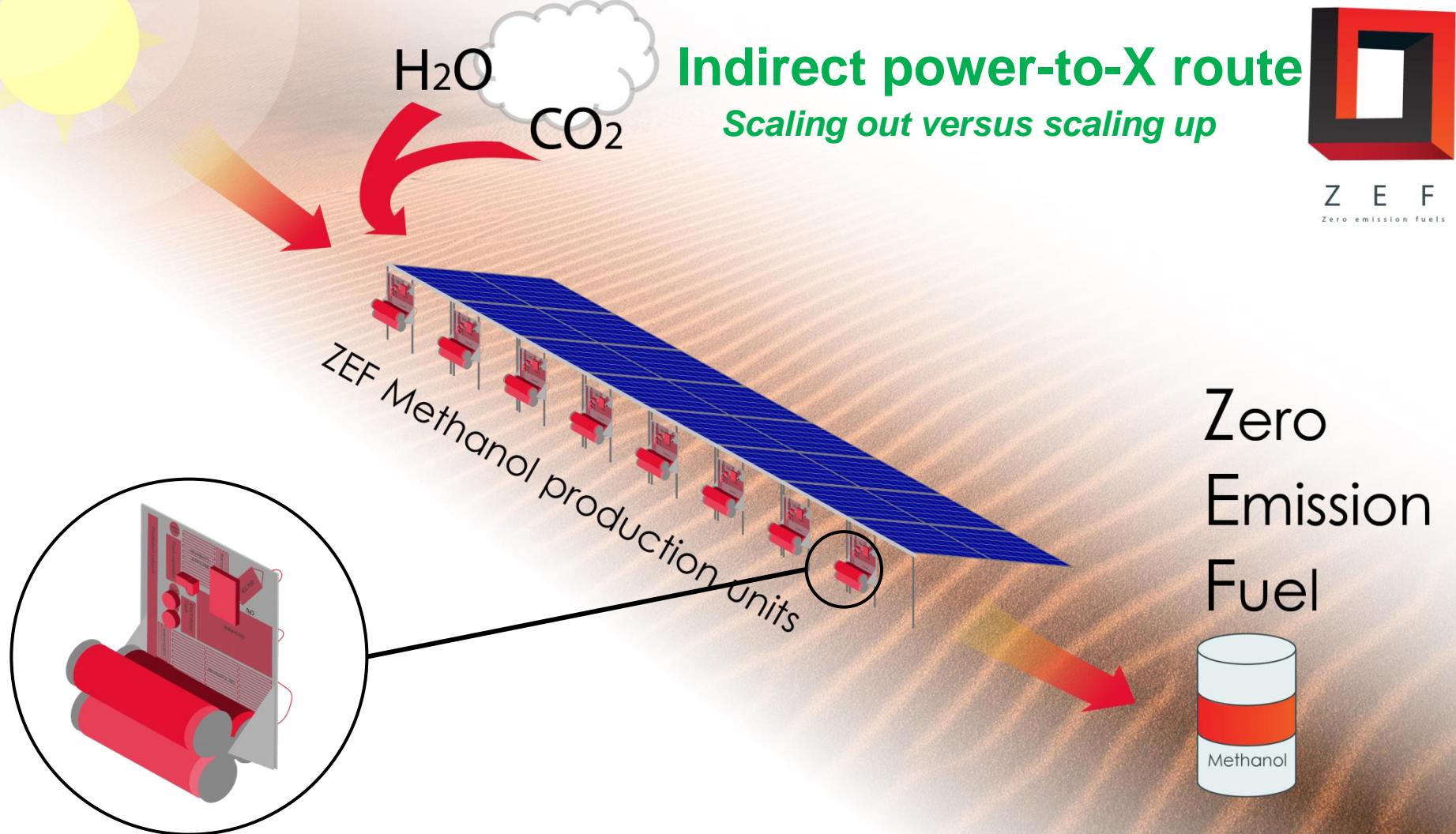


Towards an Integrated novel process



Challenge: Smart reactor system design including **Circulating Fluidized Bed** Reactor with optimal heat integration to regenerate sorbent and to make use of exothermal reaction







Global Methanol production

Methanol produced from gas (80%)

Production cost: 100–300 €/tonne

10 suppliers @ 50% marketshare

70 large production plants globally

Methanol produced from coal (17%)

Production cost: 200 – 350 €/tonne

Mainly in China

Bio Methanol (< 1%)

Production cost: 500 €/tonne

Premium sales price: 700 €/tonne

0.18 million tonnes / year

EU directive incentive: double counting in blending

IHS US GULF AND WESTERN EUROPE METHANOL CONTRACT PRICES AND
METHANEX ASIAN POSTED CONTRACT PRICE (APCP)
JANUARY 2006 - JANUARY 2017



Grey average: 350 €/tonne

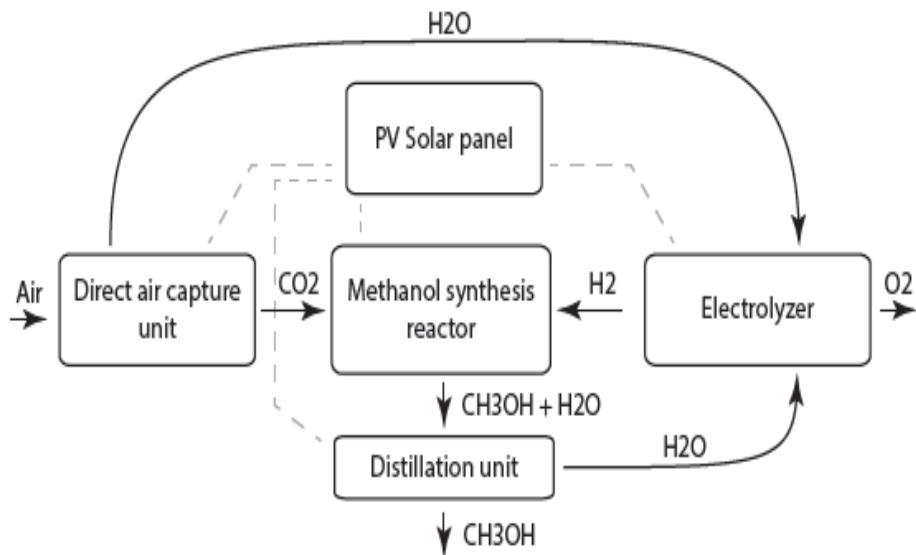
Green premium: 700 €/tonne





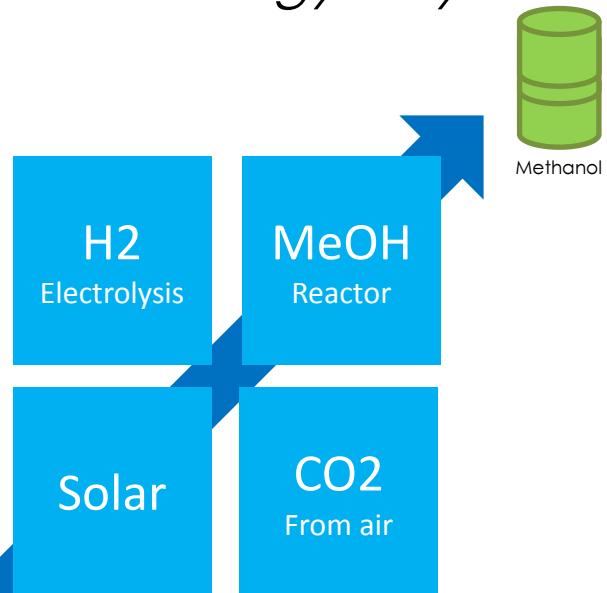
The ZEF pathway

“Capture CO₂ from air to produce renewable liquid fuels using solar energy only!”



Symbols:

- System unit
- Material stream
- Energy stream



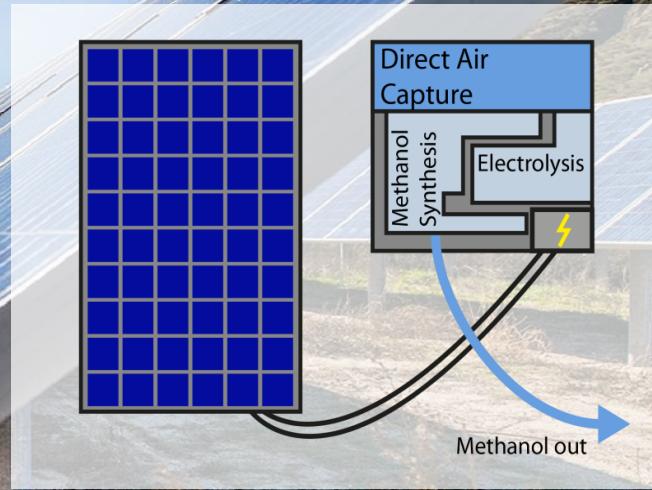
ZEF Micro-plant: numbering up



Dynamic
operation

100K Micro-plants
< 150 Components

Fast(er)
realization



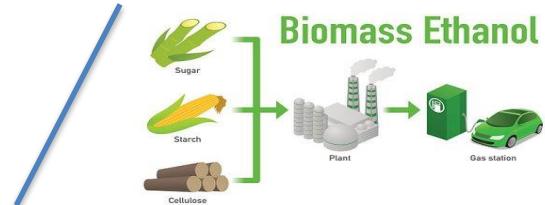


Scenario's for future production of chemicals and fuels

'Conventional'



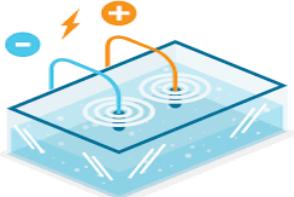
⊕



'Direct route'



⊕

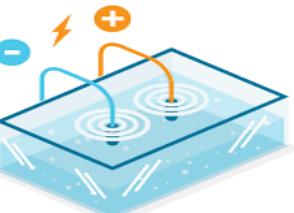


e-Refinery

'Indirect route'



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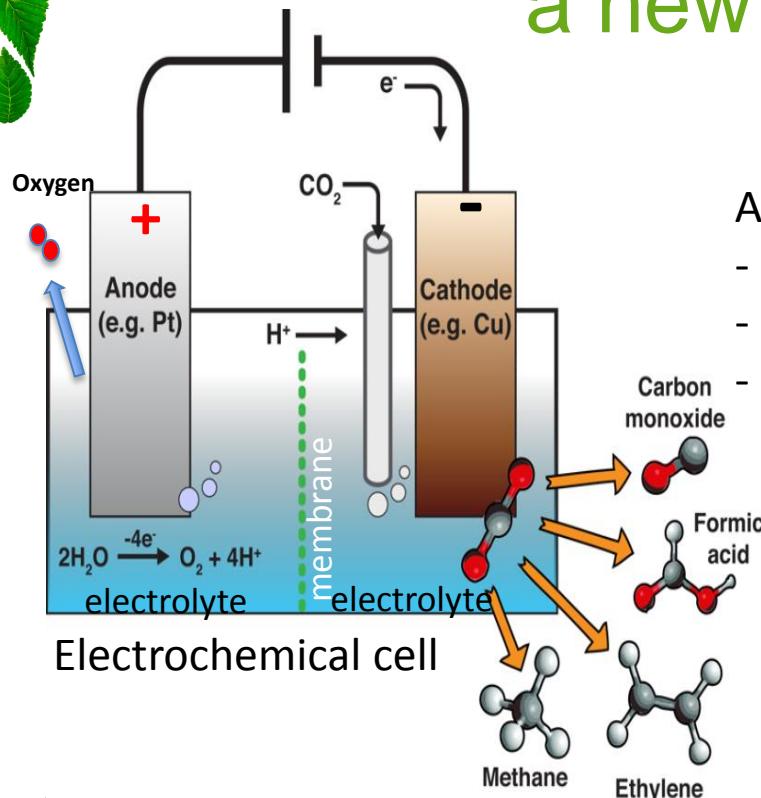


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Direct electrochemical CO₂ conversion, a new research area



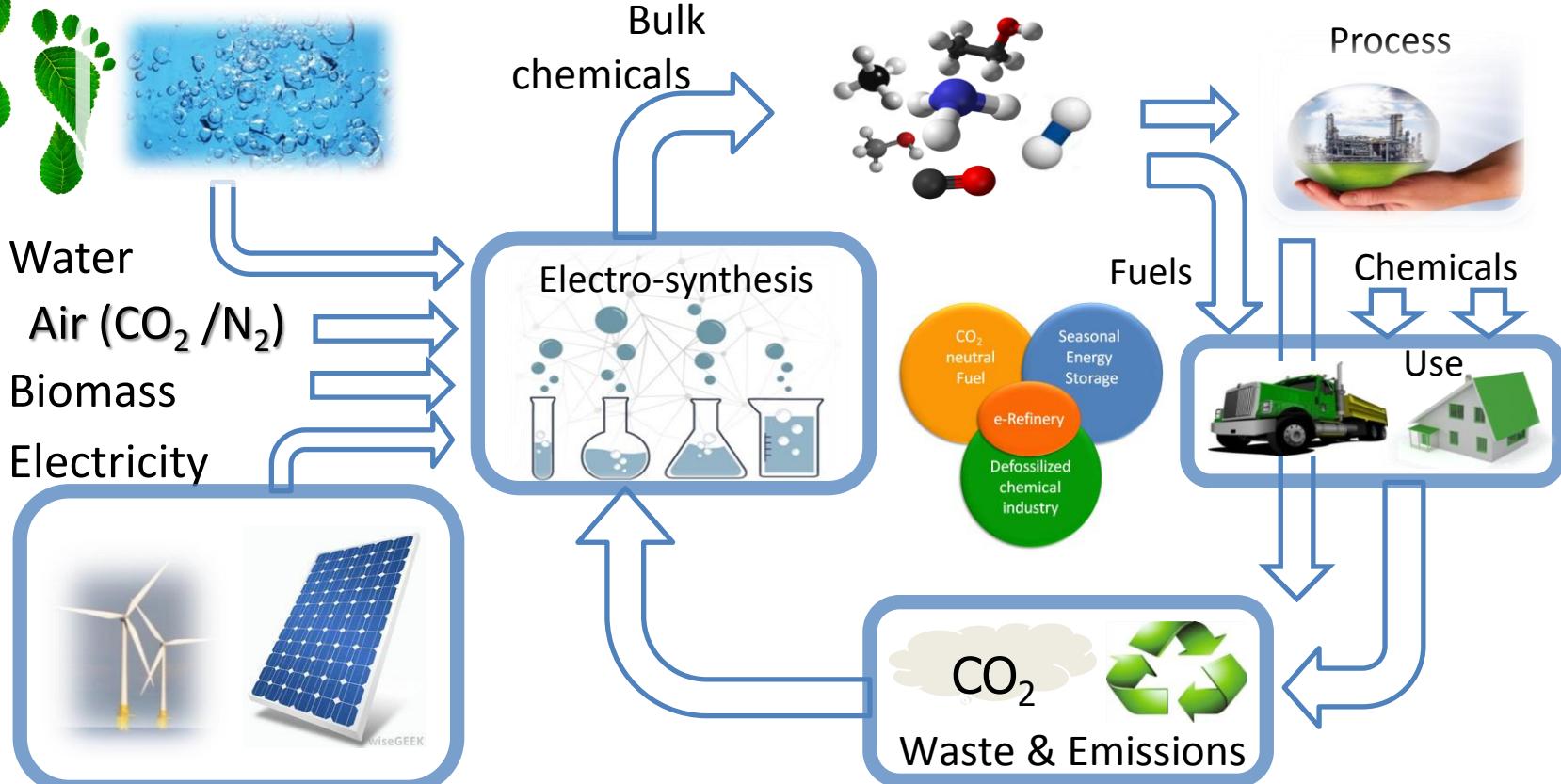
Advantages of using electrochemical conversion:

- Using DC electricity
- Mild process conditions
- Can be carried out relatively selectively

$$\Delta V = \Delta E + \sum \eta + IR$$

Equilibrium Potential Overpotentials Ohmic (resistance) loss
(thermodynamics)

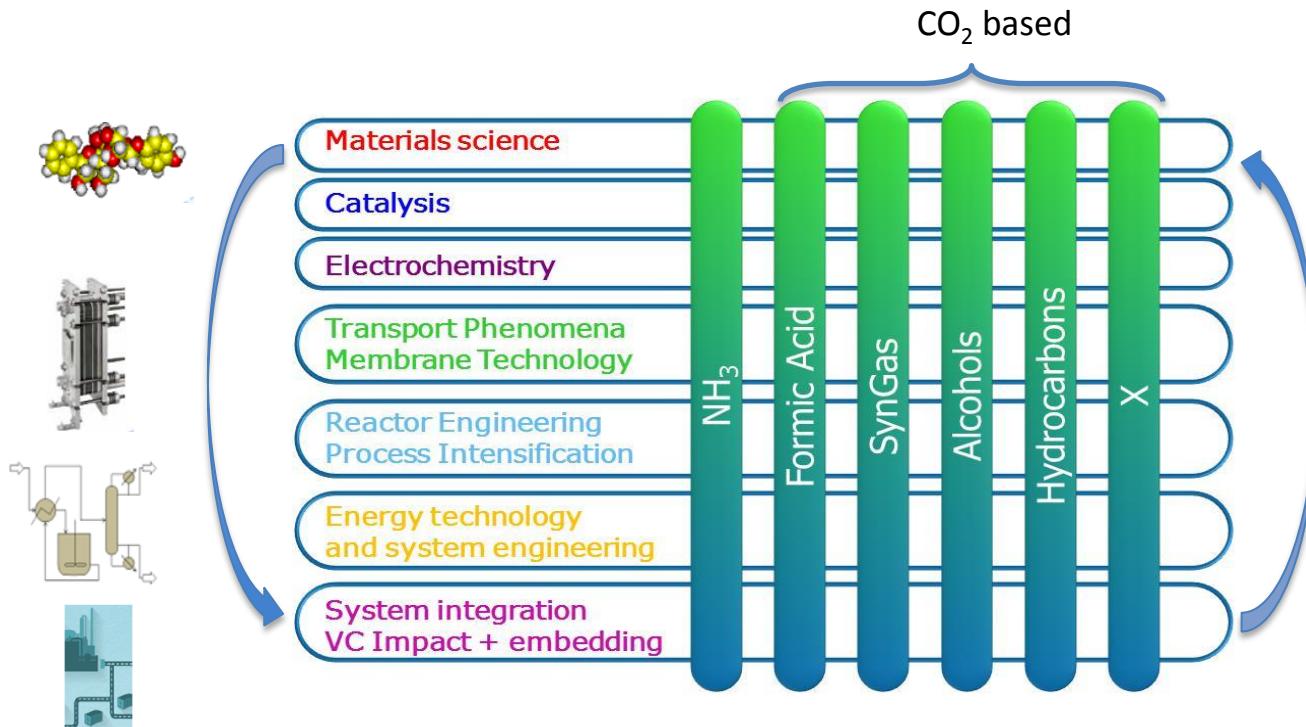
Delft Initiative: “The e-Refinery”





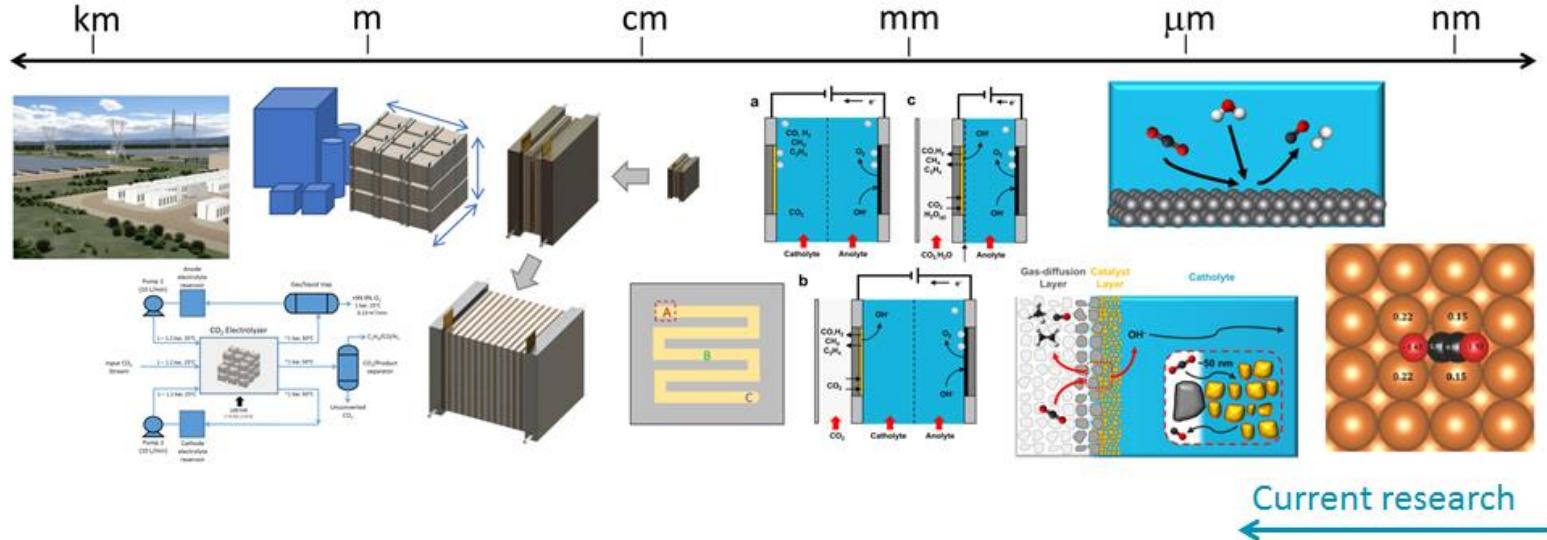
e-Refinery development at TU Delft

Fuels, bulk chemicals



Need for Speed: parallel development
together with our industry!

Bridging scales

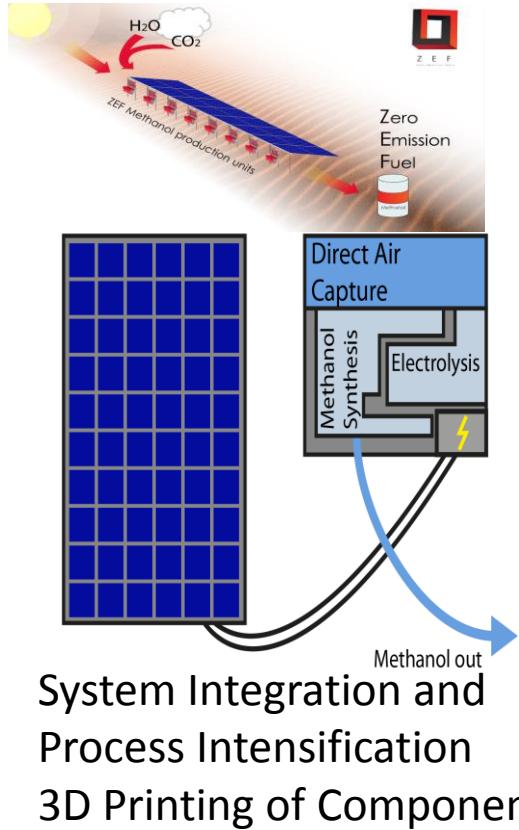




Bridging TRL levels: idea to pilots



Electrochemical conversion
Reactor CO_2 to formic acid



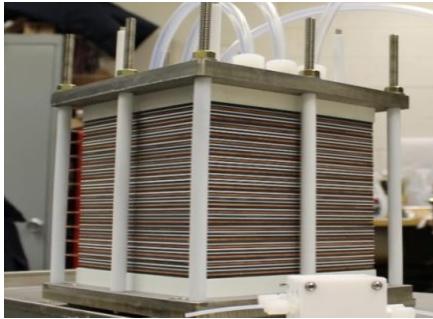
Validation



Ambition: 100 kW electrolyser for ethylene production by 2025

Operating parameters

| | |
|---|---|
| Total amperage: | ~31 kA |
| CO ₂ converted: | ~143 kg CO ₂ /day |
| C ₂ H ₄ Produced: | ~45 kg C ₂ H ₄ /day |
| H ₂ O consumption: | ~58 kg H ₂ O/day |
| O ₂ produced at anode: | ~156 kg O ₂ /day |



World production
Ethylene tending to
~200 Mton/yr
-> ~1200 GW_e needed



Goals

- Feasibility test
- Find design rules
- Basis for large scale design
- Provide input for LCA



Further developments

'Indirect route' of energy storage

- Improving fuel generation selectivity, via material development (catalyst-sorbent)
- Process integration (heat utilization, reaction-regeneration, system dynamics)
- Scale-up (process simulation & valorization, component development and testing)

'Direct route' of energy storage

- Improving, scale-up and implementing gas diffusion electrodes (increase i , lower potential)
- Design of better, stable membranes (high recovery of products)
- Design of stable electrodes (high stability in a range of pH values)
- In-situ separation of products (eliminate downstream steps)
- Use of non-aqueous solvents (easier product recovery)
- Optimize gas and liquid flows (increase concentration)
- Paired electrolysis (anode side alternative products, improved economics of overall process)



Thank you for your attention!

Acknowledgment to current projects at TU Delft:



E2CB NWO Perspectief



Toegepaste en
Technische Wetenschappen

TKI-Shell 'ToELS' project

